

**Biohydrogen from anaerobic treatment of brewery  
wastewaters: A preliminary study**

**Biohidrógeno a partir de la fermentación oscura de las  
aguas residuales de la industria cervecera: evaluación  
preliminar**

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## Abstract

Hydrogen gas is considered the energy of the future because it is renewable, clean, and does not generate greenhouse gas emissions. Among the existing processes for producing hydrogen, it is found biological production from residual wastewater with a high content of carbohydrates treated with an anaerobic process. The utilization of wastewater for hydrogen production is considered an economical alternative because the raw material is abundant and easy to access. In this sense, the present study assessed the hydrogen production from the wastewater of the beer industry treated by Up-flow Anaerobic Sludge Blanket Reactors (UASB). Then, it was applied a thermal shock (90 °C/15 min) to the inoculum, with a previous dark fermentation treatment. The results revealed that with an organic volumetric load of 9.1 gDQO/L.D., pH of 5.5, and a hydraulic retention time (HRT) of 2 h, the maximum percentage of hydrogen gas was 21 % which corresponds to a yield of hydrogen of 6.50 mMolH<sub>2</sub>/l. According to the results was found that the initial compound of wastewater influenced the stability and continuity of hydrogen gas (H<sub>2</sub>). Finally, this study identified the uncommon presence of methane gas (CH<sub>4</sub>) with hydrogen gas, despite keeping pH values nearby to 5.5, and has applied a thermal shock to the inoculum.

**Keywords:** Clean energy, dark fermentation, hydrogen, industrial wastewater.

## Resumen

El gas hidrógeno se considera como la energía del futuro por ser renovable, limpia y no generar emisiones de gases de efecto invernadero como producto final de su combustión. Entre los procesos que existen para producir hidrógeno se encuentra la producción biológica a partir de aguas residuales con altos contenidos de carbohidratos mediante procesos anaerobios. La utilización del agua residual como fuente de biomasa para producir hidrógeno se considera una alternativa económica debido a que la materia prima es abundante y de fácil acceso. Con base en lo anterior, el presente estudio evaluó la producción de hidrógeno a partir de aguas residuales de la industria cervecera tratadas en un reactor anaerobio de manto de lodos y flujo ascendente (UASB) a escala de laboratorio. Para esto, se aplicó un choque térmico (90 °C/15 min) al inóculo, para inhibir a los microorganismos metanogénicos y promover la producción de H<sub>2</sub>. Los resultados mostraron que con un valor de carga orgánica volumétrica de 9.1 gDQO/L.D., pH de 5.5 y un tiempo de detención hidráulica (TDH) de 2 h, el valor máximo de contenido porcentual de hidrógeno en el biogás fue de 21 %, lo que correspondió a un rendimiento de 6.50 mMolH<sub>2</sub>/l. A partir de los resultados, se encontró que la variación en la composición inicial del agua residual influyó en la estabilidad y continuidad de la producción del gas hidrógeno (H<sub>2</sub>). Por último, este estudio encontró la presencia de gas metano (CH<sub>4</sub>) junto con el gas hidrógeno, a pesar de mantener los valores de pH próximos a 5.5, y haber aplicado el choque térmico al inóculo.

**Palabras clave:** agua residual industrial, energía limpia, fermentación oscura, hidrógeno.

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## Introduction

Currently, 90 % of the world's energy is generated from fossil fuels, which are considered scarce and harmful to the environment and human beings. Among the alternative fuels potentially suitable for replacing fossil fuels is hydrogen. Hydrogen is a clean fuel that generates water as the only product during combustion and has 2.75 times more potential energy than hydrocarbons (Wei, Liu, & Zhang, 2010). Recently, the production of biohydrogen from different organic waste has become a low-cost and eco-friendly alternative. According to Zhong, Stevens, and Hansen (2015), at the First World Hydrogen Energy Conference held in 2006, hydrogen was identified as the first clean energy carrier of the future, not only because of its high energy density (143 MJ/kg) which is 2.6 times higher than methane and 3.3 times higher than gasoline, but also because its combustion only generates water as a by-product. Sivagurunathan, Sen, and Lin (2015) said that hydrogen production from dark fermentation has gained much attention at present mainly because there

is a wide range of substrates that can be converted to hydrogen without additional energy expenditure. Among the substrates being studied most are industrial wastewaters with high carbohydrate content.

However, biogas is not the only product of anaerobic digestion that has energy value, the biological conversion of carbohydrate-rich substrates generates intermediates such as volatile fatty acids (VFAs) which in turn are constituents of other high value-added products such as polyhydroxyalkanoates (PHA), bio alcohols such as ethanol, butanol and hydrogen gas (*i.e.*, by-products of dark fermentation). The production of hydrogen and other intermediates occurs through the dark fermentation process. Strategies to ensure effective production of hydrogen and organic acids include ensuring that the system has essentially hydrogen-producing organisms and optimizing operational conditions to ensure that this production is continuous and stable (Barca, Soric, Ranava, Giudici-Ortoni, & Ferrasse, 2015). The physiological differences between acidogenic bacteria and methanogenic archaea represent the fundamental basis for developing mechanisms to accumulate hydrogen and organic acids. A review of the literature shows that one of the most common strategies is the elimination of methanogenic organisms (Castelló *et al.*, 2020).

The brewing process involves a large number of batch processes in the processing of raw materials into the final product. Large quantities of water are used for beer production, as well as for general washing of floors, cellars, packaging, and on-site cleaning. These waters discharge a significant volume of wastewater, with high concentrations of organic pollutants characterized by high biochemical oxygen demand (BOD), chemical oxygen demand (COD), and total suspended solids (TSS)

(Arantes *et al.*, 2019). The most commonly used treatments are generally those that use anaerobic processes due to the high organic load and high biodegradability of the water (Arreola-Vargas, Flores-Larios, González-Álvarez, Corona-González, & Méndez-Acosta, 2016; Montiel-Corona & Razo-Flores, 2018). This technology in combination with aerobic processes such as activated sludge will not only allow the system to comply with discharge regulations but also generates the option of taking advantage of energy production in the form of biogas rich in methane (CH<sub>4</sub>) or hydrogen (H<sub>2</sub>). The production of CH<sub>4</sub> by anaerobic digestion (AD) in this type of water is an established and widely known technology, while the biological production of H<sub>2</sub> has not yet been widely studied (Arantes *et al.*, 2019).

There are several studies on hydrogen and methane production from both single and two-stage anaerobic processes using wastewater from the alcoholic beverage industry. Arreola-Vargas *et al.* (2016) evaluated energy recovery using the two-phase and one-phase anaerobic digestion process from the treatment of water from the tequila industry. Their results showed that using the two-phase methane production was 3.3 times higher than when operating in a single phase. Sinbuathong, Somjit, and Leungprasert (2015) treated real water from the brewery industry in 0.12 l anaerobic batch reactors to produce hydrogen. They varied the pH between 4 and 7 and applied three different types of pretreatments to the inoculum, i.e., heat shock acidification, and chloroform addition. It was mainly found that the methane-producing archaea in the sludge from UASB of the brewery tolerated the temperature change, in addition to the pretreatments by chemical additives, which led to the fact that there is no total inhibition of these

microorganisms, therefore, these authors indicated that the brewery wastewater and sludge were suitable for producing methane instead of hydrogen.

It is important to mention that one of the biggest disadvantages of biohydrogen production by dark fermentation is the presence of different organisms, consumers, and/or inhibitors of hydrogen gas. (Sikora, Błaszczuk, Jurkowski, & Zielenkiewicz, 2013). Bundhoo y Mohee (2016) indicate that there are various groups of inhibitors, among which are: H<sub>2</sub>-consuming bacteria (homoacetogenic bacteria, propionic acid fermenting organisms, methanogenic archaea), organisms that compete for the same substrate (e.g., lactic acid bacteria), and can also occur due to the presence of by-products of the same fermentation (e.g., acid accumulation). In this sense, to avoid the presence of these bacteria and increase the production of H<sub>2</sub>, it is important to develop methods to inhibit the activity of these microorganisms. Following Castelló *et al.* (2020), the procedure that has shown the best results is the application of pretreatments to the inoculum. In this sense, the objective of this study was to evaluate the production of biohydrogen through dark fermentation by applying a heat shock pretreatment (90°C/15min) to the inoculum in an up-flow anaerobic sludge blanket reactor (UASB) at a laboratory scale treating wastewater from a brewery industry.

## Materials y methods

## Wastewater

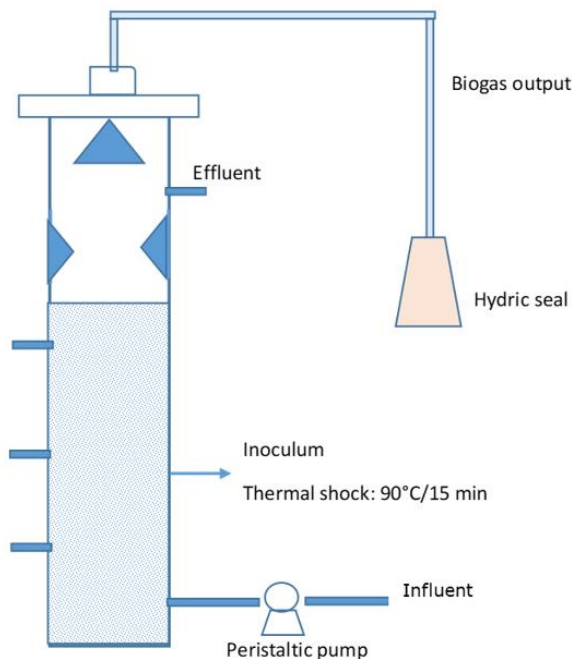
The wastewater comes from a brewery located in the city of Bogota. The samples of wastewater were collected once a week for a year from the Parshall flume located at the entrance of the Wastewater Treatment Plant (WWTP). Once the samples were collected, they were preserved and stored for later analysis. For water characterization, analyses such as total suspended solids (TSS), total dissolved solids (TDS), total solids (TS), total alkalinity, total volatile acids (TVA), sulfates ( $\text{SO}_4^{2-}$ ), chemical oxygen demand (COD), pH, total nitrogen, and phosphates ( $\text{PO}_4^{3-}$ ) were performed using the HACH model DR5000 spectrophotometer and following the recommendations of APHA (2012).

## Experimental set-up

A laboratory-scale UASB reactor was used, built-in acrylic, with a diameter (D) of 50 mm and a height (A) of 590 mm, and a useful volume of 940



ml, which resulted in a length to diameter (L/D) ratio of 12 (Figure 1) by the recommendations of Chaparro et al. (2010). The conventional UASB reactor was operated continuously for 62 days with a hydraulic retention time (HRT) of 2 h. This HRT value was chosen based on what was suggested by authors such as Preethi et al., (2019) and an average volumetric organic load of 9.41 gDQO/L.D. The inoculum was obtained from the UASB reactor that treats wastewater from the brewery industry. A heat shock treatment was applied to this inoculum at 90°C for 15 min, following the recommendations of Lay *et al.* (2019), and Wang, Fang, Fang, and Bu (2010). The pH value was adjusted to 5.5 with 1N sodium hydroxide or hydrochloric acid (10 mol/l), as appropriate.



**Figure 1.** Experimental set-up and photograph, reactor UASB. Own elaboration.

## **Analytical determinations**

To evaluate the performance of the UASB reactor, the following analyses were performed: chemical oxygen demand, total volatile acids, total alkalinity, partial alkalinity, intermediate alkalinity, total solids, suspended solids, total volatile solids, suspended volatile solids, and pH<sub>i</sub> under the methodology described in APHA (2012), and Ripley and Boyle (1986). All determinations were carried out three times a week in duplicate. The composition of the biogas, including hydrogen gas and methane gas, was analyzed in the Chromatograph brand Agilent model 7890A, equipped with a thermal conductivity detector with helium as carrier gas and a Carboxen 1010 Plot column. The temperature of the injector and the detector were maintained at 200 and 230 °C, respectively. The flow of the carrier gas was 1.5 ml/min. The volume of the sample injected was 0.7 ml.

## **Results and discussion**

Wastewater from the brewery industry is generated from several activities during production, such as cleaning of equipment and facilities, refrigeration and boiler circuits, packaging, sanitation, and factory consumption in general. As shown in Table 1, these wastewaters are characterized by high variability in most of the parameters measured. One explanation for this result may be that the sample was taken in the Parshall Flume at the entrance of the WWTP, just before the homogenizer.

**Table 1.** Characteristics of wastewater from the brewing industry.

Parameter	Unit	Median	±	D.S.*
COD	mg/l	868	±	174
pH	--	7	±	1.1
Total Alkalinity	mg CaCO <sub>3</sub> /l	1 628	±	492
TVA	mg HAc/l	1 720		317
TS	mg/l	4 915	±	1 279
TSS	mg/l	356	±	40
SO <sub>4</sub> <sup>-2</sup>	mg SO <sub>4</sub> <sup>-2</sup> /l	97	±	61
PO <sub>4</sub> <sup>-2</sup>	mg PO <sub>4</sub> <sup>-2</sup> /l	76	±	43
Cl	mg Cl <sup>-1</sup> /l	144	±	37
TN	mg TN/l	32	±	10

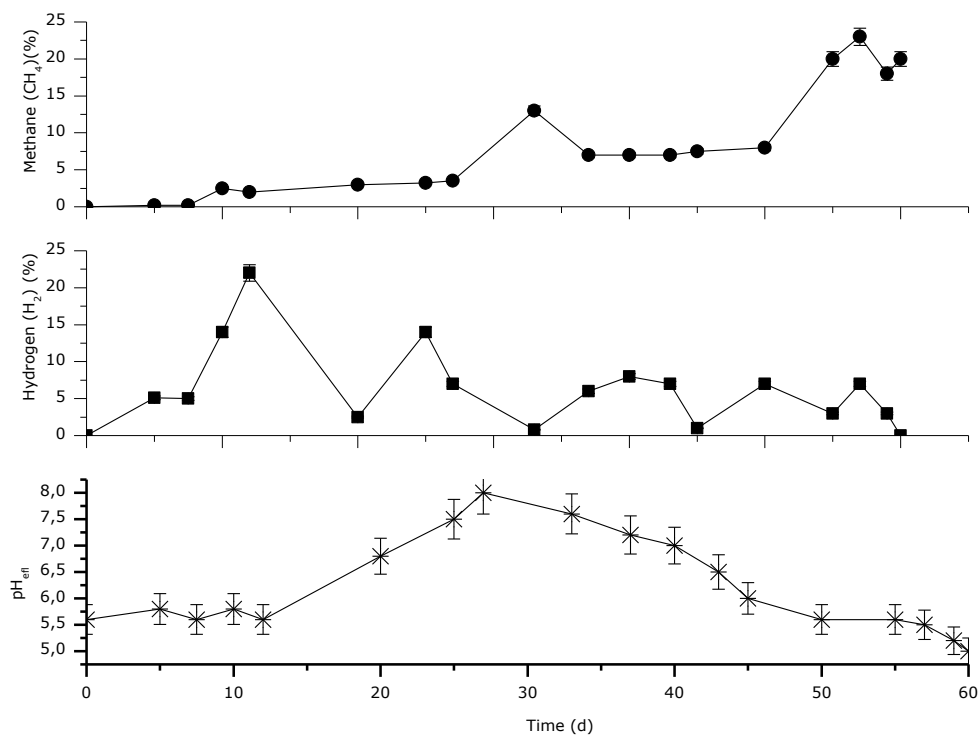
\*S.D.: Standard Deviation. Own elaboration.

From the results in Table 1, it can be said that the composition of the water could favor the production of hydrogen by anaerobic means. Additionally, the sulfate toxicity was evaluated with the COD/SO<sub>4</sub><sup>-2</sup> ratio, which resulted in an average value of 12.2. By (Chernicharo (2007) a COD/SO<sub>4</sub><sup>-2</sup> ratio greater than 10 does not cause toxicity problems due to the presence of sulfates in the anaerobic process. Finally, it was observed that the nitrogen and phosphorus content was high, mainly because of the yeast contributions to the wastewater (Gugulethu-Erogan, 2014).

## Hydrogen production

Figure 2 shows that the percentage production of hydrogen was not constant and did not reach a steady state. The steady-state is considered to be reached when the H<sub>2</sub> production does not present a variation of about 10 % as suggested by Tapia-Venegas *et al.* (2015). The maximum yield achieved was 6.50 mMolH<sub>2</sub>/l, which corresponded to 21 % of the biogas composition for operating day 12. Hydrogen production using real industrial wastewater as feedstock has shown favorable results in different studies, but under very different operating conditions, which does not allow establishing a pattern or defining values for design and operating parameters on a real scale. For example, Amorim, Alves, Martins and Amorim (2014), using an anaerobic fluidized bed reactor

treating wastewater from cassava extraction with an HRT value of 2 h produced 1.2 molH<sub>2</sub>/mol glucose. Pachiega *et al.* (2019) also studied hydrogen production from brewery wastewater, adjusting the pH value to 5.5, in anaerobic Batch reactors but with an HRT of 180 h; the maximum production was 15 molH<sub>2</sub>/mol fructose. Yu, Zhu, Hu and Zhang (2002) treated wastewater from rice production under thermophilic conditions at pH 5.5 and HRT of 2 h and their maximum yield was 2.14 molH<sub>2</sub>/mol hexose.



**Figure 2.** Temporal variation of hydrogen production, methane production, and pH in the effluent in the UASB reactor. Own elaboration.

Note: The values of the percentage production of hydrogen and methane are the result of three measurements made internally by the gas chromatograph.

Hydrogen production was unstable over time but continuous; this result can be attributed to different factors, including changes in the bacterial communities, the present consumers and/or inhibitors of H<sub>2</sub>, the reactor configuration, or the chosen hydraulic detention time. Castelló *et al.* (2020), in a wide critical review of factors influencing the stability of hydrogen production, stated that there is as yet no consensus on operational parameter values and reactor configurations for producing hydrogen from real industrial wastewater. Several studies on hydrogen production from anaerobic digestion, particularly operating in the dark fermentation phase in waters of industrial origin, noted that from approximately the 25th day of operation, there is a tendency for hydrogen production to decrease and not return to initial values (Anzola-Rojas & Zaiat, 2014; Moureira, Lima, Kmiecik, & Zaiat, 2013; Penteado, Lazaro, Sakamoto, & Zaiat, 2013).

Notwithstanding the above, some findings can be established in this study, first, H<sub>2</sub> production was not only not stable, but also did not recover over time. Approximately from day 20 of operation, it decreased until it stopped completely around day 60. This observation was similar to other authors such as Bundhoo and Mohee (2016); Sivagurunathan, Anburajan, Kumar, and Kim(2016), and Si *et al.* (2015). Particularly, Si *et al.* (2015) state that one of the causes leading to the decrease in hydrogen production is the presence of homoacetanogenic microorganisms.

Inoculum preparation plays a key role in the continuous and stable production of hydrogen, authors like Bundhoo y Mohee (2015) did an extensive review on the effect of various pretreatments on hydrogen production from dark fermentation. Among their main findings these authors indicated that, among the technologies reviewed for inoculum pretreatment, heat and acid pretreatments are the most studied and most effective. In addition, they noted that further research is needed to determine optimal pretreatment conditions before definitive conclusions can be drawn.

It is important to mention that in this study in every sample of biogas and biohydrogen was a presence of Methane gas ( $\text{CH}_4$ ) in small proportions (Figure 2). A possible explanation for this phenomenon is that, despite the thermal pretreatment of the inoculum, it was not sufficient to inhibit or eliminate the methanogenic archaea. Despite the adjustment of the pH value to 5.5 in the reactor effluent, the effluent showed average values of 6.21, except for the first 8 days of operation, when values close to 8.0 were reached, similar results were obtained by Hawkes, Hussy, Kyazze, Dinsdale y Hawkes (2007), y Palomo-Briones *et al.* (2019).

## **Variation of total volatile acids**

Anaerobic fermentation is always accompanied by the production of volatile acids. One aspect to take into account in this study is that on several occasions the wastewater already had high concentrations of total volatile acids, indicating that the water was already partially fermented. Therefore, the values were initially higher than  $> 1\ 500\ \text{mgHAc/l}$ , and on average during the observation time the increase was not higher than 20 %, as a result, an average value of  $1\ 800\ \text{mgHAc/l}$  was reached, which indicated that there was no internal production of VTA as stated by Chernicharo (2007).

Hydrogen production can be inhibited by  $\text{H}_2$  itself and by volatile fatty acids produced during the fermentation process. Fermentation with acetic and butyric acid produces 4 moles of  $\text{H}_2$  per mole of glucose and 2 moles of  $\text{H}_2$  per mole of glucose, respectively. These are the main metabolic pathways by which  $\text{H}_2$  is produced in dark fermentation Sikora *et al.* (2013). This indicates that to obtain a high  $\text{H}_2$  production, this should correlate with high acid production, a situation that does not always occur, mainly due to the way the acid is in the medium. Both acetic acid and butyric acid are weak acids that will depend on the pH value, in this sense, if the acid does not dissociate it is more likely to enter the cytoplasm of the microorganisms, causing an imbalance in the pH value, acidifying the medium and increasing the energy requirements. As a result, an imbalance in the process will be generated, and for this reason, authors like Lee, Chua, Yeoh y Ngoh (2014), Mamimin *et al.* (2015), y Ghimire *et al.* (2015) proposed that one of the best strategies to produce hydrogen from wastewater with significant variations in organic load, pH values in the acid range mainly originated in the industry, should implement the anaerobic process in two stages. In this way, the operating



parameters such as HRT, and pH, among others, are optimized in each of them, mainly favoring the microorganisms involved in each of these stages.

Finally, it is relevant to note that the production of hydrogen from real wastewater from the brewery industry is an attractive option for industries to comply with the discharge limits established in the environmental regulations and also to recover energy through the by-products generated during the treatment of their effluents. For this, the anaerobic treatment must be carried out in two stages, the first a dark fermentation phase to optimize the production of hydrogen ( $H_2$ ), and the second a phase to produce methane gas ( $CH_4$ ). In particular, it is recommended to increase the pretreatment time with heat shock for inhibition of methanogenic archaea, 15 minutes is not enough as observed in this study, despite the low hydraulic detention time used and also the pH value of 5.5. Given the heterogeneity of the raw water, trials should be conducted under factorial experiments to find the optimal range of operational parameters.

## Conclusions

In this study, a preliminary evaluation of the feasibility of producing hydrogen from dark fermentation by treating real wastewater from a brewery industry was carried out. It was found that it is feasible to produce hydrogen with this type of effluent using a conventional anaerobic UASB reactor from dark fermentation. The maximum yield achieved was 6.50 mMolH<sub>2</sub>/l, which corresponded to 21 % of the biogas composition with a volumetric organic load value of 9.1 gCOD/L.D. However, it was not possible to maintain the stability and continuity of the process despite the application of thermal shock to the inoculum and pH control. It is important to mention that future studies must evaluate how factors such as variability in the composition of the wastewater, the configuration of the reactor, how the inoculum pretreatment is carried out, and its origin influence to improve of the stability of the biological process and consequently the continuity in the generation of hydrogen gas.

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